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SOLAR CELLS OF METAL-FREE PHTHALOCYANINE DISPERSED IN POLYVINYL CARBAZOLE

(I) EFFECTS OF THE RECRYSTALLIZATION OF H₂PC ON CELL CHARACTERISTICS

Michiko Shimura and Hideo Baba



Translation of "H₂PC-PVK maku o mochiiru taiyo denchi.

(I) H₂PC-no seisei ni yoru seru tokusei-no henka", Denki kagaku (Journal of the Electrochemical Society of Japan), Vol. 50, No. 6, 1982, pp. 678-683

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For the purpose of developing an organic semiconductor solar cell, the effects of the recrystallization of metal-free phthalocyanine (H2PC) on the characteristics of NESA/H2PC-PVK/Au sandwich cells have been in vestigated. Alfa-H2PC sandwich cells showed photovoltage and photocurrent in a H2PC direction opposite to that shown by as-supplied H2PC cells, which consisted mainly of beta-H2PC. Some difference was observed in the response times of the two cells. It was considered that photocharacteristics change with the specific resistance of the H2PC, which is related to its crystal forms. In the cells with Iow-resistance H2PC, (1.e., alfa-H2PC), carriers were generated in H2PC by illumination, while in high resistance H2PC (i.e., beta-H2PC) cells, carriers were generated in PVK which was sensitized with H2PC.				
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SOLAR CELLS OF METAL-FREE PHTHALOCYANINE DISPERSED IN POLYVINYL CARBAZOLE

(I) EFFECTS OF THE RECRYSTALLIZATION OF H₂PC ON CELL CHARACTERISTICS

M. Shimura^X), H:Baba^{X)}

1. Introduction

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Among various methods of energy conversion solar cells are the most clean and noiseless. There is a problem of cost, which, however, is reduced from year to year *1. A coefficient of conversion in the case of tandem cells can be as high as 50 and more [1]. In the last few years more and more investigations are carried out with regard to utilization of organic semiconductors for the manufacture of solar cells*2. Organic materials are not so restrictive from the point of view of resources as inorganic materials. Furthermore, their advantage is that they can be easily produced in the form of thin films for lower cost. Manufacturing of thin films is performed by dripping an organic material dispersed in a slurry-type high-molecular binder onto the surface of a substrate [4]. This method may be very promising for the future as it is free of disadvantages in rent in evaporation methods, i.e. it does not require the use of thermally decomposable materias and prevents losses due to scattering.

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^{*1} In 1970 the cost was \$500/watt, whereas in 1980 it was reduced down to \$10/watt. In 1986 the cost may be as low as \$70/watt[1].

^{*2} It has been reported that phthalocyanine [2-4], chlorophyll [5], merocyanine, hydroxy-squalilium compounds [6] are used for this purpose.

^{*} Numbers in the margin indicate pagination in the foreign text.

However, under conditions of high light intensity, solar cells incorporating organic materials reveal a disadvantage consisting of an extremely small coefficient of conversion. This stems from the low quantum efficiency of carrier generation in the case of organic materials, low mobility of carrier and high trap density [7]. Furthermore, in this case one cannot neglect the light-retaining effect caused by a metal film existing on the path of light due to application of the Schottky barier on the plane of interface.

An object of the present investigation was to measure "p-n junctionness" of an organic material per se in order to reduce its specific resistance by changing carriers. due to various additives. As a first step for investigating the effect of purification, phthalocyanine, which is known as a chemically-stable cosmetic has been chosen. Metal-free phthalocyanine (hereinafter referred to simply as H₂PC), which is advantageous from the point of view of light absorbance, was selected for experiments. Susceptibility to the effect of purification of H₂PC also has been considered in a sandwich-cell structure, in view of a weak barrier action on both sides of the interface between NESA/H₂PC/Au used as electrodes.

2. Experiment

2.1. Preparation of H₂PC-PVK Film

 ${\rm H_2PC}$ was used as supplied, by purchasing this high-purity material from Eastman Kodak or, after purification, by recrystallization from concentrated sulfuric acid. Finely divided powder of ${\rm H_2PC}$ in agate mortar (with an average size of particles equal to 47 $\mu {\rm m}$) was added to a slurry-type polyvinyl carbazole (the product of Tokyo Kaseisei, hereinafter referred to as PVK) solution (the solvent comprised monochlor benzene and dichlor ethane taken in

1:1 ratio). Apart from special cases, an amount of additives comprised about 40% by weight. Experiments which have been conducted by the authors [8] confirmed that this proportion is the most advantageous from the point of view of the photovoltaic power and stability of the film (resistance to cracking and peeling-off from the substrate). The H₂PC-PVK solution was stirred for several hours in a magnetic stirrer until uniform distribution had been reached. Following this, the solution was dripped onto the surface of the substrate, and then dried, first naturally and then by keeping it for a day in a vacuum-type desiccator.

2.2. Preparation of Sandwich Cell

NESA glass was used as a substrate ($\mathrm{Sn0}_2$ film, $\mathrm{N}_D\approx 2\times 10^{19}$ cm⁻³, a product of 'Matsuzaki Shinkusei Co.) the surface of which was covered with a film of $\mathrm{H}_2\mathrm{PC}\text{-PVK}$ (an average thickness of the film was about $48\mu\mathrm{m}$)* and then an electrode of Au (with a thickness of about 300Å) was applied onto the film by a spattering method. Lead wires from the electrode were formed by sticking copper foil pieces to one end of the Au electrode with the use of an Ag paste (a product of Du Pont, No. 4922). Tapping of the lead wires from the NESA film was performed first by vacuum depositing indium onto the end of the NESA film and then by sticking a copper foil piece onto said layer with the Ag paste.

2.3. Measurement of Photovoltaic Power and Photocurrent

In order to exclude the effect of atmospheric humidity and electric noise on the results of measurements, the latter were conducted in

^{*3} Electric-type film thickness gage (a product of Tokyo Seimitsu Sei K.K., Minicom, EDM-M306) was used for measurements

in a vacuumeter (a product of Pirex Co., 10^{-2} Torr capacity) which in turn was installed inside a metal shield box (with a small opening for the optical path). An electrometer (a product of Tageta Rikensei K.K., mod. TR-8651* 4) also was used for measuring circuit interruption voltage $V_{\rm oc}$ and short-circuit current $I_{\rm sc}$.

A 500W Xenon lamp was used as a light source and monochromatic light of various wavelength between 700 and 400 nm was selected by means of color glass filters (Toshiba Garasu K.K., Y-48) passing the light with $\lambda>480$ nm (light intensity* 5 : 32.4 mW·cm $^{-2}$)* 6 and interference filters (interference filters manufactured by Mitsunobu Kogaku K.K., λ_1 : about 10 nm; light intensity: about 0.3 mW·cm $^{-2}$).

Input impedance of the electrometer during the measurement of $V_{\rm oc}$ was $10^{14}\Omega$ and the following impedances were applied during the measurement of $I_{\rm sc}$: input impedance for the range of $10^{-11}\Lambda$ was $10^9 \sim 10^8~\Omega$; input impedance for the range of $10^{-10}\Lambda$ was $10^8 \sim 10^7~\Omega$; and input impedance for the range of $10^{-6}\Lambda$ was $10^4 \sim 10^3~\Omega$. The response time of the electrometer (the time of oscillation of the indicator until it shows 99% of the rated input signal current) for currents above $10^{-10}\Lambda$ was within the limits of 0.5 sec.

^{*&}lt;sup>5</sup> Measurements were conducted with the use of a compressor-type thermopile (a product of Eppley Co., 16-junction Bi-Ag).

 $^{*^6}$ Emission of light with $\lambda >$ 480 nm was selected in order to hinder, the light absorption by PVK and to reveal the light absorption effect from the side of ${\rm H_2PC}$ itself.

3. Results of Experiments

3.1. Variation of H, PC under Effect of Purification

Under the effect of recrystallization from concentrated sulfuric acid $\rm H_2PC$ changes its color (from blue tinged with green to blue) and increases in volume. However, X-ray diffraction patterns (Figure 1) differ only by wider peaks at 6.8~7.2, 13.5~16.7 and 24.7.27.3°. This material was established as α -

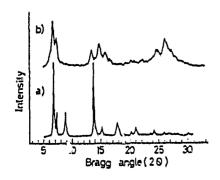


Fig.1. X-ray diffration pattern of H₂PC (a) as-supplied; (b) recrystallized.

H₂PC [9-11]. On the other hand, an unpurified material was recognized as β-H₂PC[9-11] and it was impossible to detect the impurities by means of X-ray diffraction. However, it proved to be possible to determine the absorption caused by impurities (the absorption maximum at 644 nm and shoulder at 633 nm), apart from the absorption from H₂PC (695, 657 nm) in the unpurified material, by resolving said H₂PC in monochlor benzene* and subsequently determining the visual absorption spectrum.

Figure 2 illustrates an IR absorption spectrum of $\rm H_2PC$ (KBr disk method). All absorption spectrums obtained from the purified $\rm H_2PC$ belonged to $\alpha-\rm H_2PC$, whereas almost all absorption spectra from unpurified material turned out to be $\beta-\rm H_2PC$ with a small quantity of particles in the range of 1570, 1520 and 1360 cm⁻¹, which, however, cannot belong to $\alpha-\rm H_2PC$.

Monochlor benzene was chosen as a solvent because it possesses low solubility with respect to ${\rm H_2PC}$ whereby, without interfering with the absorption of ${\rm H_2PC}$ per se it allows detection also of the absorption of impurities.

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The results given above show the way for exclusion of impurities and the manner of conversion from β to α -modification of H_2PC by means of recrystallization from concentrated sulfuric acid.

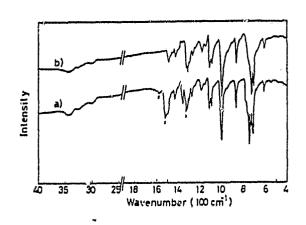


Fig.2. IR spectra of H₂PC by KBr disk method:
(a) as-supplied; (b) recrystallized

Furthermore, although it is not shown in the drawings, H_2SO_4 hardly shows any residual activity *8 (due to adsorption and formation of electric charge-transfer complexes*9) in the recrystallized material. The following description relates to the effect of purification of H_2PC when it is dispersed in a PVK film.

3.2. Photovoltaic Power

Under conditions of illumination by 480 nm light the purification

of H₂PC considerably changes the photovoltaic power generated in H₂PC-PVK film sandwich cells (Figure 3). As an ordinate in this drawing, the voltage of the NESA-pole is plotted with an Au-pole taken as a criterion. About +200 mV voltage (however, with the illumination of the Au-pole side) was obtained for purified H₂PC cells as compared to -180 mV for the unpurified H₂PC (with the illumination of the NESA-pole) in the case of measurement of a steady photovoltaic power. The same symbols are maintained with the change of the planes of illumination and only the photovoltaic power is reduced. In the

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^{*8}The absorption of SO_4^{2-} cannot be detected by means of IR.

When ESR measurements of $\rm H_2PC$ powder are carried out at room temperature, a strong signal is obtained in the vicinity of g= 2.002 [12]. Because this is not related with the recrystallization, the results are the same both for unpurified and purified $\rm H_2PC$. It has been assumed that in this case electrons are trapped

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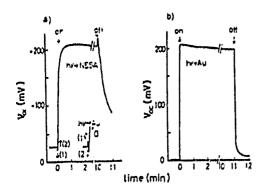


Fig.3. Typical time profiles of open circuit voltage of $H_2PC-PVK$ sandwich cells by the illumination of $\lambda > 480$ nm light:

- (a) as-supplied H₂PC;
- (b) recrystallized

determination of working spectrums with regard to the steady photovoltaic power, the absorption spectrum of a film with the illumination of the NESA-pole in an unpurified H₂PC cell and that of a film with the illumination of the Au-pole in a purified H₂PC cell are close to each other.

Complicated transient phenome'na* 10 take place in an unpurified

H2PC cell until conditions of steady
photovoltaic power are reached. In
other words, when the NESA-pole side
is illuminated, the photovoltaic power
is first shifted towards a positive
side, then is converted into a negative
value, amplified and at last reaches
a steady value (Figure 3a). Also in
the case of illumination of the Aupole side the voltage is shifted to-

wards a nagative value, then towards a positive value, then again becomes negative and reaches a steady value (Figure 3a). Detailed explanations for these phenomena are given further in Chapter 3.3.

in structural cracks of ${\rm H_2PC}$. However, at the present time this question is again under investigation. On the other hand, Calvin [13] has published the information on ${\rm H_2PC}$ and chloranil in connection with the ESR absorption due to electric charge-transfer complexes of ${\rm H_2PC}$.

 * The regeneration ability of the transient phenomena is very strong and it does not change even with cyclically repeated illumination or after heat treatment (30 min at 80°C under 10⁻² Torr).

On the other hand, in purified H₂PC cells, along with the illumination, the photovoltaic power is shifted towards a positive side and quickly reaches steady conditions (Figure 3b).

Time necessary to achieve steady conditions comprises in unpurified H2PC cells about 7 sec and in purified H2PC about 1.6 sec. Also the time required to restore the initial voltage after stopping illumination in unpurified H2PC cells is equal to 2 min and in purified H2PC is equal to 35 sec. The difference in speeds of optical response is determined mainly by difference in the usual trac density. However, it may be supposed that divergence of carries generated in the films, i.e. the question where the majority of carries takes place -- in H,PC or in PVK, stems from differences in planes of separation of charges, which has been mentioned above, and from differences in directions of generation of the photovoltaic power associated with this phenomenon. This is the final source of deviations in the form of crystals of ${\rm H}_2{\rm PC}$. According to the results of measurements of the specific resistance, in a dark room of tablet forming facilities, in the present investigation of H_2PC powder, [14] * 11 3.3 x $10^{12}\Omega$ cm was obtained for an unpurified material, i.e. for $\beta-H_2PC$, whereas a value of 1.0 x $10^7\Omega\cdot cm$ was obcained for a purified material, i.e. for $\alpha-H_2PC$. The reason for which the specific resistance is different in various modifications of crystals is known [15], however some deviations may be introduced due to difference in methods of measurements and conditions of H2PC (tablets, vacuum deposited film, monocrystal). It is supposed that A-H2PC has a small value of specific resistance due to the irregular chrystalline structure and small activation energy of dark current generation, and also due to the fact that wavelength limits in the generation of photoelectric current are longer than the wavelength in \$ -modification [16] *12. Apart from this, multiple states of the carrier may be

<sup>*11
100</sup> mg of H₂PC was treated in an IR tablet-forming machine under vacuum conditions (400 kg cm⁻²). By means of spattering Au was applied onto both sides of the thus formed tablet. Measurements were conducted by applying a stable voltage at 15°C in a vacuometer (10⁻² Torr).
*12

According to Liang [16] the activation energy of dark conduction of

caused by a disordered state leading to incompleteness in the crystal= line structure in band gaps of λ -H₂PC. Due to considerable differences in the specific resistance of various crystalline modifications, H₂PC has a smaller specific resistance than PVK. Therefore in the case of application of H₂PC with high carrier concentration, the Fermi level is determined by means of the carrier in H₂PC. On the other hand, in the case of high specific resistance, which is equal or higher than that of PVK, i.e. when H₂PC is dispersed in the film with low concentration of the carrier, the Fermi level is determined by the carrier in PVK. Thus, in this case H₂PC in the film plays the role of a sensitization pigment with respect to PVK. Such a concept is supported by the behavior of photoelectric current described in the subsequent chapters.

3.3. Short-Circuit Photovoltaic Current

The following remarkable effect of purification of $\rm H_2PC$ has been observed in measuring the short-circuit photovoltaic current $\rm I_{\rm sc,ph}$, when a sandwich cell is connected to the direct electrometer and illuminated with $\lambda > 480 \, \rm nm$. In other words, there are no special distinctions in the direction *13 and generation efficiency of current $\rm I_{\rm sc,ph}$, when, in an unpurified $\rm H_2PC$ cell, this current reaches its steady value equal to $-1.2 \times 10^{10} \, \rm A \cdot cm^{-2}$ (however, when the NESA pole side is illuminated, apart from the light penetration factor of NESA pole equal to 0.82, this value has to be conpensated with regard to the in- $\frac{681}{12}$ cident light illuminous intensity) and when, in a purified $\rm H_2PC$ cell, it reaches its steady value at $+9.7 \times 10^{-8} \, \rm A \cdot cm^{-2}$ (however, when the Au pole side is illuminated, apart from the Au pole penetration factor of 0.38, this value has to be compensated with regard to the incident light/illiminous intensity). Also, in this case, complicated transient

 H_2PC has been defermined as follows: $A-H_2PC$: 1.4eV, $\beta-H_2PC$: 1.9eV. Also the wavelen;th limits in generation of the photovoltaic current were the following: $A-H_2PC$: 0.68eV, $\beta-H_2PC$: 0.79eV.

Symbol means that current is flowing from the NESA pole to the outside circuit.

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phenomena take place. Although these phenomena are almost the same as have been described in the previous chapter with regard to photovoltaic power, the following details may be added. In the case when the NESA pole side is illuminated, the photoelectric current has \oplus sign, i.e. it flows from the NESA pole towards an outside circuit. Then after 0.1 sec the current is reversed to \ominus sign, i.e. the current $\mathbf{I}_{\text{SC},\text{ph}}$ flows yowards the NESA pole. This \ominus $\mathbf{I}_{\text{SC},\text{ph}}$ current immediately (in 0.3 sec after reversing) reaches its maximum, and following this, it is slowly reduced until the steady state is achieved. Figure 4(a) illustrates active spectra of \ominus and \ominus $\mathbf{I}_{\text{SC},\text{ph}}$ in the transient area. One can see that both are very close to the absorption spectra of the film. Therefore it can be supposed that both are formed under the effect of light absorption in $\mathbf{H}_2\text{PC}$ ajoining the NESA pole. Among them the best efficiency was

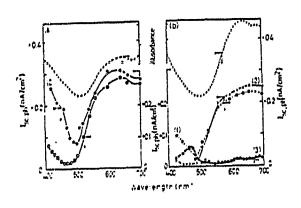


Fig. 4. Action spectra of transient short-circuit photovoltaic current of as-supplied H₂PC-PVK sandwich cell: Illuminated (a) though NESA and (b) through Au. The currents are corrected for transmittances of NESA, Au, and interference filters. The absorption spectra of (a) assupplied H₂PC in thin PVK film and (b) as-supplied H₂PC powder:....

shown (in the area of the wavelength exceeding 520 nm) by \oplus I_{sc.ph} (curve (1)) which has a very quick speed of response. Comparison of these results with the behavior of cells (curve (1)) in the case of illumination on the Au pole, side shown in Figugure 4(b), suggests that on the basis of holes formed in H,PC the Dember effect's photoelectric curgent takes place. *14. With regard · to Θ I_{sc,ph} (curve (2)), the $\mathfrak{g}e..$ neration of which is slightly delayed, it should be noted that the best efficiency occurs within the wavelength range below 500 nm. This suggests that, in the case when $\mathrm{H}_{2}\mathrm{PC}$ acts with respect to PVK as a sensitizer, one can assume that carriers are generated in PVK [18]. (Figure 4(b), cirve (3)). When the Au-pole side is illuminated in an unpurified H2PC cell, first a $ext{\'o}$

negative current Isc,ph, file. the current flowing from the Au pole towards the outer circuit, occurs, and then immediately (in 0.1 sec) after this it reverses to a Φ positive value and flows from the water circuit towards the Au-pole . In 0.5 sec the current reverses again to a (negative value which increases and at last reaches its steady value. Thus, in the case of illuminating the Au-pole side the current passes through complicated transient phenomena and its direction ... experiences complicated variations. Figure 4 (b) shows action spectra of each $I_{sc,ph}$. One can see that both, $\bigcirc I_{sc,ph}$ (curve (1)), having quick response, and $\bigcirc I_{sc,ph}$ (curve (2)), approach the absorption spectrum of β -H₂PC. It confirms the fact that these currents are generated under the effect of light absorption in H,PC adjoining to the Au-pole. The quick-responsive Θ I_{sc,ph} current flows from the illumination plane (the Au-pole) towards the outer circuit. First of taic current Isc, ph is observed flowing from the illumination plane (the NESA pole) towards the outer circuit. It is suggested that in this case the grotovoltaic current can be associated with the Dember effect of elegtron holes formed in H,PC under the action of illumination. However, it is supposed that the quick-responsive negative Θ current I sc.ph also occurs under the Dember effect.

As far as the positive \oplus current $I_{\text{sc,ph}}$ (curve (2)) is concerned, it should be noted that, referring to the behavior (Figure 5(a)) in the purified H_2PC cell, which is considered below, the photovoltaic

The following experiments were conducted to verify the existence of the Dember effect in a H_2 PC-PVK film. A sandwich cell was formed by placing a thick α - H_2 PC-PVK film between two NESA films (NESA/ α H_2 PC-PVK/NESA sandwich), and following this, the $I_{sc,ph}$ current was observed along with $V_{oc,ph}$ by illuminating one side of the sandwich with the λ > 480 nm light. As a result, generally the potential on the illuminated surface was increased and the outer circuit was formed by the photovoltaic current which flowed from the illuminated surface to the dark surface. Holes were formed in H_2 PC.

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current can be considered, with the use of the potential gradient on the interface plane between H_2PC and Au, as the motion of excited electrons in the optically excited H_2PC . The \bigcirc negative $I_{sc,ph}$ cur-

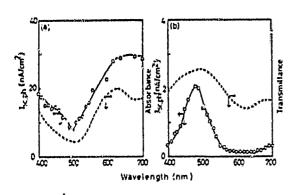


Fig. 5. Action spectra of transient short-circuit photovoltatic current of recrystallized H₂PC-PVK sandwich cells:
Illuminated (a) through Au and (b) through NESA. The currents are corrected for transmittance of NESA, Au and interference filters. The absoption (a) and transmission (b) spectra of recrystallized H₂PC in PVK film:

(curve (3)), which has rent the most delayed response, by configuration of its spectrum approaches that of the transmission spectrum of the film. Therefore, this corresponds to separation of charge on the back surface of the film (NESA-pole Its current I sc,ph has a delayed response but is generated under steady conditions. suggests that in cells produced on the basis of an unpurified H2PC the electric chargerseparation dominates on the boundary surface of the NESA pole, and current flows from the film towards the NESA pole, i.e. the negative \bigcirc I_{sc.ph} current flows under steady conditions. Consideration of the interface plane between H2PC and PVK shows that in view of high concentration of carriers (as in the case of the present investigation), the NESA-

pole has a Fermi level of about 4.9eV [4]. On the other hand, with regard to PVK, it was revealed that its ionization potential is equal to 6.1 eV, and electronic affinity takes place at 1.5 eV. Therefore with a relatively high concentration of carriers, as in the present investigation, the Fermi level exists near the band of valence electrons. For this reason a barrier is formed on the interface plane* 15,

A similar barrier is formed on the interface with Au. Therefore the energy band of PVK decreases on both boundary planes of NESA and Au. and a protrusion should be formed due to the bulk activity. However, in reality

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It should be noted with regard to this interface plane that if the illumination performed excites PVK per se, then the transfer of electrons starts in the direction from the excited PVK towards the NESA-pole and photoelectric current is generated (I_{sc,ph} has a negative Θ value because it flows from the outside circuit towards the NESA-pole). Furthermore, when illumination is carried out by means of light having a long wavelength, the light is absorbed by H₂PC adjoining PVK. Perhaps the photovoltaic current is generated due to charge transfer, when the excitation level of H₂PC and that in PVC have superimposed wave functions.

There have been many reports [18, 19] on the photoconductivity effect of PVK, when sensitivity was increased by means of a pigment. Okado et al. 19 \ast^{16} observed the initial spike in a system composed of a pigment applied by means of vacuum deposition onto the surface of a PVK film. The phenomenon took place under the effect of a charge transfer between the PVK and pigment. The conclusion can be drawn on the basis of the above results, that in cells produced with the use of an unpurified 12 PC, the photovoltaic current is induced by carriers in PVK and that 12 PC acts as a sensitivity increasing pigment for PVK.

Action spectra shown in Figure 5 were obtained for cells manufactured on the basis of a purified $\rm H_2PC$. In the case of illumination on the Au-pole side, these spectra approach the absorption spectrum of α -H₂PC (Figure 5(a)) and in the case of illumination on the side of the NESA pole, the configuration approaches the transmission spectrum of α -H₂PC (Figure 5(b)). This clarifies the fact that the plane of separation of the electric charge in a purified H₂PC cell is a boundary plane between H₂PC and Au and that the photovoltaic current is generated due to excitation of $\rm H_2PC.*^{17}$

the charge separation on the boundary surface of NESA is dominant. At the present time this phenomena is under investigation.
*16

The initial spike of the photovoltaic current is induced by electrons, in the case of an n-type pigment, and by traps formed by holes in PVK, in the case of a p-type pigment.

A similar behavior has been observed by Sharp [4] in a NESA/ $X-H_2PC-PVA/Al$ sandwich cell. However, in this case the Schottky barier is formed on the interface between $X-H_2PC$ (E_F : about 5.09 eV⁴) and Al (about 4.2eV⁴), which has a low work function. In the present investigation an apparent barrier of this type appears in the case of conjugation with Au having a high work function *18. However this is a topic of the next report.

3.4. Behavior of PVK Film Sensitized by H₂PC

The previous chapter described observation of behavior in the case of pigment sensitization, with regard to PVK and the use of unpurified H₂PC. Probably this is related to the high specific resistance of unpurified H₂PC and low mobility and high concentration of carriers in H₂PC.

In this chapter a subject of discussion comprises behavior of $\sqrt{683}$ PVK sensitized with H_2 PC, in the case of α - H_2 PC of low specific resistance dispersed in small quantities in PVK.

The results showed that, when the quantity of $\alpha-H_2PC$ in the film is less than 5% by weight, the behavior is the same as in cells produced with unpurified H_2PC . In other words, the current $I_{sc,ph}$ changes from the outer circuit towards the NESA-pole, and the action spectrum shows that the charge separation plane is transferred to the NESA/film boundary plane (Fig. 6). The speed of optical response also is delayed, behind the value of that in unpurified H_2PC cells. This confirms the previously mentioned facts about sensitizing action in unpurified H_2PC cells. With regard to H_2PC , the carrier concentration depends essentially on the type of crystals. With high concentration

^{*1.7} Typical rectifying characteristics (the direction is assumed as forward when the NESA pole is polarized as a positive one) were obtained in determining V-I relationship on the dark side with application of voltage to the NESA/purified $\rm H_2PC-PVK/Au$ cell from the outer side. On the other hand, the rectifying char existics were definitely absent in cells based on unpurified $\rm H_2PC$.

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'tration of carriers, p-type occurs, whereas low concentration produces an intrinsic type, both having different Fermi levels. Thus, a barrier of a certain type is formed on the interface plane, which: suggests that it can act as a charge separation plane. However, this

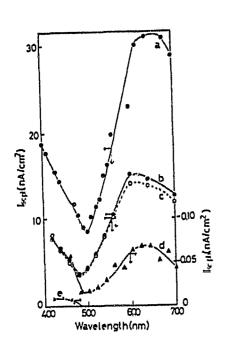


Fig. 6 Action spectra of short-circuit photovoltaic current of α-H₂PC-PVK sandwich cells, concentrations of α-H₂PC are 40(a), 20(b), 15(c),5(d), and 0 wt%(e): a,b and c were illiminated through Au; d and e were through NESA. The currents are corrected for transmittance of NESA, Au and interference filters.

question is still under investigation.

Figure 6 illustrates action spectra,
when a sandwich cell is made only of
a PVK film. For PVK used in the
cells, optical characteristics are shown
from the minimum values to 460 nm.

These characteristics were analyzed on
the basis of excitation correlated with
the level of impurities in PVK.

Thus, the present experiment clarifies an effect of impurities contained in unpurified H₂PC.

4. Conclusions

The following conclusions can be drawn on the basis of investigations of the effect of H₂PC purification on characteristics of a SA/H₂PC-PVK/Au cell, the investigation being conducted for the purpose of developing semiconductor so car cells:

(1) Cells formed by dispersing in a PVK film, $\alpha-H_2$ PC, obtained by recrystallization from concentrated sulfuric acid, along with the photovoltaic

^{*18}

The same phenomenon has been observed when, apart from Au, other materials such as Ag and In were used as poles.

power, which is opposite in its direction to that in an unpurified H_2PC cell, produce a photoelectric current (the main component of unpurified H_2PC is $\beta-H_2PC$). The speed of optical response also is different.

- (2) It is suggested that the distinctions of cell characteristics mentioned above stem from differences in specific resistances, which in turn depend on the type of H_2PC crystals. In the case when $\alpha-H_2PC$ of low specific resistance is used, carriers in a film approach holes of H_2PC , and when $\beta-H_2PC$ is used, they approach holes in PVK. In the last case, H_2PC acts as a sensitizing pigment for PVK.
- (3) In order to improve characteristics of cells it is necessary either to reduce resistance of H₂PC or to choose another high molecular binder apart from PVK.

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